

## COUMARINS FROM UNRIPE FRUITS OF *PONCIRUS TRIFOLIATA*

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**Key Word Index**—*Poncirus trifoliata*; Rutaceae; coumarins; poncimarín; isoponcimarín.

**Abstract**—Petrol extracts of unripe fruits of *Poncirus trifoliata* L. were found to contain 7-geranyloxy coumarin, bergapten, imperatorin, 6-methoxy-7-geranyloxy coumarin and two new coumarins which were shown by chemical and spectroscopic means to be 7-(3'-methyl-2',3'-epoxybutyloxy)-8-(3"-methyl-2",3"-epoxybutyl) coumarin and 7-(3'-methyl-2',3'-epoxybutyloxy)-8-(3"-methyl-2"-oxobutyl) coumarin respectively.

### INTRODUCTION

During a current project dealing with the coumarin derivatives of *Poncirus trifoliata* L., an investigation of the whole unripe fruit demonstrated that it is a very rich source of coumarins but not furocoumarins, as opposed to seeds from ripe fruit in which furocoumarins largely predominate. Previously the furocoumarins imperatorin, bergapten, isopimpinellin, prangenin and prangenin hydrate, and a small quantity of the two coumarins 7-geranyloxy coumarin (aurapten) and 6-methoxy-7-geranyloxy coumarin were isolated from seeds [1, 2]. This paper describes the isolation from a petrol extract of unripe fruit of two new isoprenylated coumarinic derivatives, which we called with the trivial name poncimarín (5) and isoponcimarín (6), as well as a large amount of the above mentioned coumarins.

A number of coumarins and alkaloids of rutaceous plants are known in which isopentenyl and geranyl side chains are present. These are at various oxidation levels and the question is raised if the oxidation levels of such terpenoid residues can be correlated with their taxonomic distribution [3]. The new extractives were identified as 7, 8 disubstituted coumarin [4, 5], with two oxidised isopentenyl chains linked to an O atom in the 7 position and to a C atom in the 8 position respectively, and can be regarded as osthenol (7-hydroxy-8-(3'-methyl-2'-butenyl) coumarin derivatives.

Brown *et al.* [6] established that osthenol and 7-demethylsuberosin (7-hydroxy-6-(3'-methyl-2'-butenyl) coumarin are intermediates in angular and linear furocoumarin biosynthesis respectively, osthenol being much less rapidly metabolized than 7-demethylsuberosin. Therefore it is possible that poncimarín and isoponcimarín also may be implicated in the angular furocoumarin or pyranocoumarin pathways and from this point of view it is significant that these compounds, which are well represented in unripe fruits, are not detectable even by TLC in extracts of ripe fruits or seeds. Up to now however, only the angular pyranocoumarin seselin has been isolated from *P. trifoliata* roots [7], while in fruit neither derivatives of the angelicine nor seselin type have been detected.

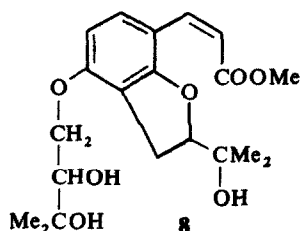
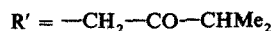
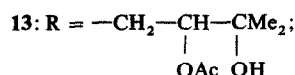
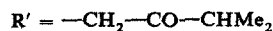
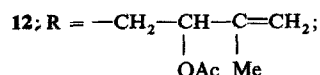
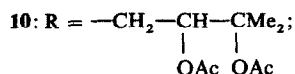
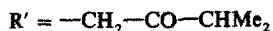
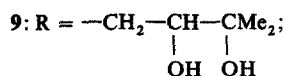
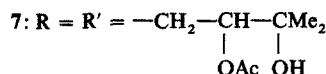
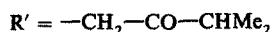
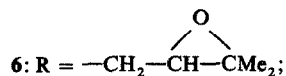
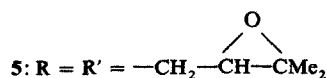
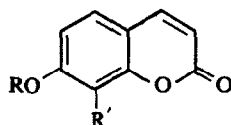
### RESULTS

As described in the Experimental Section, the isolation of the various coumarinic and furocoumarinic compounds was achieved by extraction of the unripe fruits of *P. trifoliata* with petrol and by subsequent chromatography of the extract on a silica gel column.

The first compound eluted from the column was 7-geranyloxy coumarin (1) (aurapten), which has been found in many rutaceous plants and identified by acid hydrolysis to umbelliferone and further methylation to herniarin and by comparison of physical and spectral data (UV, IR, PMR) with those of an authentic sample. In the next fractions a mixture was eluted of the two furocoumarins bergapten (2) and imperatorin (3), which were separated by further column chromatography. Further elution of the column gave 6-methoxy-7-geranyloxy coumarin (4) identified from spectral data and by acid hydrolysis to scopoletin. Up till now, this coumarin has been found in *Feronia elephantum* [8], *Haplaophyllum hispanicum* (*H. linifolium*) [9] and *H. pedicellatum* [10] in the Rutaceae.

In subsequent fractions two compounds were eluted which had a small difference in  $R_f$  values. The combined oily residues were dissolved in EtOAc and after long standing furnished a substantial quantity of a solid, which crystallized from MeOH and gave the first new coumarin, poncimarín (5). The compound showed a blue fluorescence in UV light,  $C_{15}H_{22}O_5$  from elemental analysis and MW determination, typical UV spectrum of a 7-alkoxycoumarin, very similar to that of osthenol and auraptenol [11]. The PMR spectrum of 5 showed doublets centered at  $\delta$  6.26 and 7.65 assigned to the coumarinic system protons at H-3 and H-4 and doublets centered at  $\delta$  6.9 and 7.37 for H-6 and H-5 *ortho* aromatic protons respectively. A 4 line signal centered at  $\delta$  4.23 was assigned to the O-methylene. A broad multiplet from  $\delta$  3 to 3.4 was assigned to the benzylic methylene and to the two epoxide protons. Four singlets at  $\delta$  1.30; 1.39; 1.41; 1.51 occurred for the two gem-diMe groups.

Acid hydrolysis carried out by adding a small quantity of conc  $H_2SO_4$  to a HOAc solution of 5 gave a mixture of products, the major of which was purified by column



chromatography and identified as the *bis*-acetate 7. It would be expected to obtain a monoacetate of the 2',3' diol derivative of osthenol and/or ( $\pm$ ) lomatin (3'-hydroxy-3',4'-dihydroseselin) derived from an acid catalysed cyclization [12] and/or dihydrooselone, the dehydration derivative of lomatin [13] from acid hydrolysis of 5, but none of the hydrolysis products showed a  $R_f$  value corresponding to that of synthetic samples of the anticipated compounds. The unexpected resistance to acid hydrolysis of the C-7 ethereal linkage in poncimarín was later confirmed under the same conditions with isoponcimarín (6) and with synthetic 7-(3'-methyl-2',3'-epoxybutyloxy) coumarin.

Confirmation of the branching from the 8 position of the coumarin skeleton of an epoxide isopentenyl residue followed from treatment of 5 with methanolic KOH, when the compound 8 was formed; the structure of this compound was deduced from PMR analysis. Such transformation, well known in the coumarin series [14], involves the opening of the lactone ring followed by intramolecular reaction between the phenolic and the epoxide groups forming a new ring. Un-

equivocal confirmation of the structure 5 for poncimarín followed from a synthesis achieved by isoprenylation of osthenol and PBA epoxidation of the isoprenylated derivative [15].

Rechromatography of the combined residue from which poncimarín was obtained gave the second new coumarin, isoponcimarín (6). It showed a blue fluorescence in UV light,  $C_{19}H_{22}O_5$ , from elemental analysis and MW determination, a UV spectrum virtually superimposable on that of poncimarín and typical therefore of a 7-alkoxycoumarin chromophore. The PMR spectrum of 6 confirmed the presence of a 7,8 disubstituted coumarin system showing doublets centered at  $\delta$  6.23 and 7.64 for the H-3 and H-4 protons and doublets centered at  $\delta$  6.87 and 7.38 for the H-6 and H-5 *ortho* aromatic protons respectively. A sharp singlet at  $\delta$  4.07 was assigned to the benzylic protons of the C-8 branched isoprenylic moiety, because of the combined diamagnetic effects of the aromatic nucleus and contiguous carbonyl group [11], while a septet at  $\delta$  2.86 and doublet at 1.23 were assigned to the isopropyl group of the same chain. A 4 line signal centered at  $\delta$  4.19 and a triplet centered at  $\delta$  3.08 was

assigned to *O*-methylene and to epoxide proton respectively, while two singlets at  $\delta$  1.35 and 1.37 were assigned to the terminal Mes.

Isoponcimarins are not an artefact from the rearrangement of an epoxide group of poncimarins, since the latter compound did not undergo isomerization under the condition of isolation and the crude extracts showed the presence of both coumarins on TLC. Isoponcimarins by treatment with  $(\text{COOH})_2$  in boiling  $\text{H}_2\text{O}$  underwent opening of the epoxide ring to give the diol 9, which by acetylation gave the diacetate 10.

Like poncimarins, the etheral linkage also appeared to be resistant to acid hydrolysis. Under these conditions the epoxide group of 6 underwent rearrangement giving the diketone derivative 11 and acetolysis to the monoacetate 13 of the diol 9, which by elimination of  $\text{H}_2\text{O}$  gave the compound 12.

The downfield shift of the methyne proton of the monoacetate 13 by comparison with the diol 9 confirmed the proposed structure in accordance with Krassusky rule [16], with attachment of the acetyl group to the carbon atom bearing the greater number of free H atoms.

#### EXPERIMENTAL

PMR spectra were recorded at 60 MHz unless otherwise specified and are given in  $\delta$  (Hz) relative to TMS; all assignments are in agreement with relative peak areas and where applicable with decoupling experiments. UV spectra were obtained in EtOH and IR spectra as KBr discs. Mps, determined in open capillary, are uncorr.

**Isolation.** Fruits of *P. trifoliata* were collected from plants growing in Padua or its surroundings in June, ca 3 months before ripeness. They were dried at  $60^\circ$  to constant wt (5.85 kg), finely ground and exhaustively extracted with petrol. Solvent was removed and the oily residue (572.5 g) chromatographed on 5%  $\text{H}_2\text{O}$  deactivated Si gel. Fractions from the column were monitored by Si gel TLC using cyclohexane-EtOAc (13:7). Initial petrol fractions contained trace amounts of a compound with a violet fluorescence ( $R_f$  0.9) which was not isolated. The next fractions, eluted with petrol- $\text{C}_6\text{H}_6$  (17:3), contained a blue fluorescent compound ( $R_f$  0.75) and yielded 1 (19.5 g), mp  $71^\circ$ , from *n*-hexane. Found: C, 76.39; H, 7.37. Calc. for  $\text{C}_{19}\text{H}_{22}\text{O}_3$ : C, 76.48; H, 7.43%. Hydrolysis of 1 in HOAc soln by addition of few drops of conc  $\text{H}_2\text{SO}_4$  gave umbelliferone (7-hydroxycoumarin) which by treatment with  $\text{CH}_2\text{N}_2$  in  $\text{Et}_2\text{O}$  gave herniarin (7-methoxycoumarin). UV, IR and PMR spectra were identical with those of an authentic specimen. Petrol- $\text{C}_6\text{H}_6$  (7:3) fractions containing two yellow fluorescent components ( $R_f$  0.57 and 0.52) gave after rechromatography a small amount of bergapten (2) and imperatorin (3). Workup of the subsequent  $\text{C}_6\text{H}_6$ -EtOAc (9:1) eluate gave 6-methoxy-7-geranyloxycoumarin (4) (1.6 g), mp  $88^\circ$ , after recrystallization from EtOH  $95^\circ$ . Found: C, 73.19; H, 7.28; —OMe, 9.49. Calc. for  $\text{C}_{19}\text{H}_{24}\text{O}_4$ : C, 73.14; H, 7.37; —OMe, 9.44%. Acid hydrolysis of 4 in HOAc soln gave scopoletin (6-methoxy-7-hydroxycoumarin) identical to a synthetic specimen. Fractions eluted with  $\text{C}_6\text{H}_6$ -EtOAc (17:3) contained 2 compounds ( $R_f$  0.25 and 0.23) both with a pale blue fluorescence. Solvent was removed and the residue dissolved in EtOAc. After a long standing poncimarins (5) crystallized (5.7 g), mp  $140^\circ$ , after recrystallization from MeOH,  $[\alpha]_D^{25} -56.2^\circ$  ( $\text{CHCl}_3$ );  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ) 320 (4.19), 255.5 (3.61), 246 (3.6), 235 (3.57), 216 (shoulder; 4.14),  $\lambda_{\text{min}}$  262 (3.28), 250.5 (3.55), 240.5 (3.54), 233 (3.56);  $\nu$  1725, 1600, 1500, 1390, 1250, 830; PMR ( $\text{CDCl}_3$ , 90 MHz)  $\delta$  7.65 (1H, *d*,  $J = 9.5$  Hz) H-4; 7.37 (1H, *d*,  $J = 8.6$  Hz) H-5; 6.9 (1H, *d*,  $J = 8.6$  Hz) H-6; 6.26 (1H, *d*,  $J = 9.5$  Hz) H-3; 4.23 (2H, 4 line signal) — $\text{OCH}_2$ —; 3.4–3 (4H, *m*) epoxide ring protons and benzylic methylene; 1.51; 1.41; 1.39; 1.3 (each 3H, *s*) C-Me. Found: C, 69.10; H, 6.89.  $\text{C}_{19}\text{H}_{22}\text{O}_3$  requires: C, 69.07; H, 6.71%. Workup of the mother

liquors by Si gel chromatography gave a further crop of crystalline 5 (2.3 g) and isoponcimarins (6) (7.04 g), mp  $85^\circ$ , from *n*-hexane,  $[\alpha]_D^{20} -6.94$  ( $\text{CHCl}_3$ ),  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ) 320 (4.20); 255 (3.63); 217 (shoulder 4.16),  $\lambda_{\text{min}}$  264 (3.38);  $\nu$  1720, 1600, 1500, 1395, 1375, 1280, 1250, 1110, 1050, 875, 830, 760; PMR ( $\text{CDCl}_3$ )  $\delta$  7.64 (1H, *d*,  $J = 9.5$  Hz) H-4; 7.38 (1H, *d*,  $J = 8.6$  Hz) H-5; 6.87 (1H, *d*,  $J = 8.6$  Hz) H-6; 6.23 (1H, *d*,  $J = 9.5$  Hz) H-3; 4.19 (2H, 4 line signal) — $\text{OCH}_2$ —; 4.07 (2H, *s*)  $\phi$ — $\text{CH}_2$ —CO—; 3.08 (1H, *t*) epoxide ring proton; 2.86 (1H, septet,  $J = 7$  Hz) and 1.23 (6H, *d*,  $J = 7$  Hz) — $\text{CHMe}_2$ ; 1.37 and 1.35 (each 3H, *s*) C-Me. Found: C, 69.01; H, 6.71.  $\text{C}_{19}\text{H}_{22}\text{O}_3$  requires: C, 69.07; H, 6.71%.

**Acid hydrolysis of poncimarins.** To a soln of 5 (540 mg) in HOAc (5 ml) 5 drops of conc  $\text{H}_2\text{SO}_4$  were added and the mixture kept at room temp. for 4 hr. The mixture was diluted with  $\text{H}_2\text{O}$ , neutralized with  $\text{NaHCO}_3$  and extracted with EtOAc. After removal of solvent, chromatography of the syrupy residue led to isolation of the major from several products. This compound, which failed to crystallize, was the diacetate 7, PMR ( $\text{CDCl}_3$ )  $\delta$  7.62 (1H, *d*,  $J = 10$  Hz) H-4; 7.32 (1H, *d*,  $J = 8.6$  Hz) H-5; 6.76 (1H, *d*,  $J = 8.6$  Hz) H-6; 6.15 (1H, *d*,  $J = 10$  Hz) H-3; 5.45–4.9 (2H, *bm*) two — $\text{CH}(\text{OAc})$ —groups; 4.4–4.15 (2H, *m*) — $\text{OCH}_2$ —; 3.3–3.05 (2H, *m*) benzylic protons; 2.0 (6H, *s*) acetyl groups; 1.43 (12H, *bs*) C-Me. Found: C, 61.68; H, 6.68.  $\text{C}_{23}\text{H}_{30}\text{O}_9$  requires: C, 61.32; H, 6.71%.

**Alkaline hydrolysis of poncimarins.** 5 (1.035 g) was dissolved in MeOH (20 ml) and a soln of methanolic KOH (15 ml) added. The mixture was allowed to stand 18 hr at room temp.  $\text{H}_2\text{O}$  added and the MeOH removed. The aq. soln was acidified with dil  $\text{H}_2\text{SO}_4$  and extracted with  $\text{Et}_2\text{O}$ , the organic phase washed with  $\text{NaHCO}_3$  soln and the solvent evaporated. From the residue the main component was isolated by Si gel chromatography and identified as 8 uncrystallizable, PMR ( $\text{CDCl}_3$ ; 90 MHz)  $\delta$  7.45 (1H, *d*,  $J = 8.5$  Hz) and 6.42 (1H, *d*,  $J = 8.5$  Hz) *ortho* benzenic protons; 6.94 (1H, *d*,  $J = 12.5$  Hz) and 5.89 (1H, *d*,  $J = 12.5$  Hz)  $\beta$  and  $\alpha$  protons respectively of the *cis*-coumarinate system; 4.64 (1H, *t*,  $J = 9$  Hz)  $\alpha$  proton of the dihydro furan ring; 4.22–4.07 (2H, 4 line signal) — $\text{OCH}_2$ —; 3.72 (3H, *s*) —OMe; 3.27–3.03 (3H, *m*)  $\beta$  protons of the dihydro furan ring and — $\text{CH}(\text{OH})$ —; 2.9–2.4 (*brs*; displaced by  $\text{D}_2\text{O}$  addition) —OH; 1.40, 1.36, 1.34, 1.20 (each 3H, *s*) C-Me. Found: C, 62.78; H, 7.48.  $\text{C}_{20}\text{H}_{28}\text{O}_7$  requires: C, 63.14; H, 7.42%.

**Hydrolysis of isoponcimarins.** (a) A  $\text{H}_2\text{O}$  soln of 6 (300 mg) and  $(\text{CO}_2\text{H})_2$  (300 mg) was boiled for 10 min and after cooling extracted with EtOAc. After removal of solvent, the residue was purified by Si gel chromatography and elution with EtOAc. The pure diol 9, which could not be crystallized, had PMR ( $\text{CDCl}_3$ )  $\delta$  7.62 (1H, *d*,  $J = 9.5$  Hz) H-4; 7.33 (1H, *d*,  $J = 9$  Hz) H-5; 6.85 (1H, *d*,  $J = 9$  Hz) H-6; 6.2 (1H, *d*,  $J = 9.5$  Hz) H-3; 4.45–3.9 (2H, *m*) — $\text{OCH}_2$ —; 4.03 (2H, *s*)  $\phi$ — $\text{CH}_2$ —CO—; 3.95–3.65 (1H, *t*, being each peak split into doublets) — $\text{CH}(\text{OH})$ —; 3.5–2.9 (*ca* 2H, *brs*; displaced by  $\text{D}_2\text{O}$  addition) —OH; 2.83 (1H, septet,  $J = 9$  Hz) and 1.19 (6H, *d*,  $J = 7$  Hz) — $\text{CHMe}_2$ ; 1.25 (6H, *s*) C-Me. Found: C, 65.78; H, 7.02.  $\text{C}_{19}\text{H}_{24}\text{O}_6$  requires: C, 65.50; H, 6.94. The diol 9 was refluxed with  $\text{Ac}_2\text{O}$  and few drops of  $\text{C}_5\text{H}_5\text{N}$ .  $\text{H}_2\text{O}$  was added and after cooling, the mixture was neutralized with  $\text{NaHCO}_3$  and extracted with EtOAc. After removal of solvent the diacetate 10 was crystallized from cyclohexane, mp  $118.5$ – $119^\circ$ , PMR ( $\text{CDCl}_3$ )  $\delta$  7.57 (1H, *d*,  $J = 9.5$  Hz) H-4; 7.31 (1H, *d*,  $J = 9$  Hz) H-5; 6.75 (1H, *d*,  $J = 9$  Hz) H-6; 6.14 (1H, *d*,  $J = 9.5$  Hz) H-3; 5.6–5.35 (1H, 4 line signal) — $\text{CH}(\text{OAc})$ —; 4.30–4.50 (2H, *m*) — $\text{OCH}_2$ —; 3.95 (2H, *s*)  $\phi$ — $\text{CH}_2$ —CO—; 2.92 (1H, septet,  $J = 7$  Hz) and 1.24 (6H, *d*,  $J = 7$  Hz) — $\text{CHMe}_2$ ; 2.08 and 2 (3H, each; *s*) acetyl groups; 1.55 (6H, *s*) C-Me. Found: C, 63.81; H, 6.55.  $\text{C}_{23}\text{H}_{28}\text{O}_8$  requires: C, 63.88; H, 6.53%. (b) A soln of 6 (580 mg) in HOAc (2 ml) was added to conc  $\text{H}_2\text{SO}_4$  (0.15 ml) and kept at room temp. for 4 hr. After a further addition of conc  $\text{H}_2\text{SO}_4$  (0.15 ml) the mixture was allowed to stand 2 hr, and then diluted with  $\text{H}_2\text{O}$ , neutralized by addition of a  $\text{NaHCO}_3$  soln and extracted with  $\text{Et}_2\text{O}$ . The residue (626 mg) obtained from the dried  $\text{Et}_2\text{O}$  extract was chromatographed on anhydrous Si gel and eluted with  $\text{CHCl}_3$ . Two compounds

( $R_f$  0.22 and 0.31) were eluted in the first fraction. The residue was dissolved in  $\text{CHCl}_3$ , *n*-hexane added and the mixture cooled in a solid  $\text{CO}_2$ - $\text{Me}_2\text{CO}$  bath. A white ppt. of 11 was collected and twice crystallized from  $\text{CHCl}_3$ -*n*-hexane (1:20) (52.2 mg), mp 109°, PMR ( $\text{CDCl}_3$ )  $\delta$  7.60 (1H, *d*,  $J = 9$  Hz) H-4; 7.34 (1H, *d*,  $J = 8.5$  Hz) H-5; 6.67 (1H, *d*,  $J = 8.5$  Hz) H-6; 6.21 (1H, *d*,  $J = 9$  Hz) H-3; 4.72 (2H, *s*)  $-\text{OCH}_2\text{CO}-$ ; 4.1 (2H, *s*)  $\phi-\text{CH}_2-\text{CO}-$ ; 2.85, 2.8 (1H each, septets,  $J = 7$  Hz) and 1.2, 1.1 (6H, each, *d*,  $J = 7$  Hz)  $-\text{CHMe}_2$  groups. Found: C, 68.69; H, 6.61.  $\text{C}_{19}\text{H}_{22}\text{O}_3$  requires: C, 69.07; H, 6.71. By rechromatography of the mother liquors from 11, pure 12 was obtained but it could not be crystallized. PMR ( $\text{CDCl}_3$ ):  $\delta$  7.63 (1H, *d*,  $J = 9$  Hz) H-4; 7.37 (1H, *d*,  $J = 8$  Hz) H-5; 6.73 (1H, *d*,  $J = 8$  Hz) H-6; 6.21 (1H, *d*,  $J = 9$  Hz) H-3; 5.56 (1H, *t*,  $J = 6$  Hz)  $-\text{CH}(\text{OAc})-$ ; 5-5.18 (2H, *m*)  $-\text{CMe}=\text{CH}_2-$ ; 4.18 (2H, *d*,  $J = 6$  Hz)  $-\text{O}-\text{CH}_2-$ ; 4.01 (2H, *s*)  $\phi-\text{CH}_2-\text{CO}-$ ; 2.83 (1H, septet,  $J = 7$  Hz) and 1.24 (6H, *d*,  $J = 7$  Hz)  $-\text{CHMe}_2$ ; 2.1 (3H, *s*) acetyl group; 1.83 (3H, *brs*)  $-\text{C}(\text{=CH}_2)\text{Me}$ . Found: C, 67.48; H, 6.61.  $\text{C}_{21}\text{H}_{24}\text{O}_6$  requires: C, 67.73; H, 6.50%. Further elution of the column of the rearrangement mixture gives 13 (171 mg) mp 104° from  $\text{EtOAc}$ -*n*-hexane, PMR ( $\text{CDCl}_3$ )  $\delta$  7.56 (1H, *d*,  $J = 9$  Hz) H-4; 7.29 (1H, *d*,  $J = 8.5$  Hz) H-5; 6.78 (1H, *d*,  $J = 8.5$  Hz) H-6; 6.15 (1H, *d*,  $J = 9$  Hz) H-3; 5.3-5.05 (1H, 4 line signal)  $-\text{CH}(\text{OAc})-$ ; 4.5-4.1 (2H, *m*)  $-\text{O}-\text{CH}_2-$ ; 3.94 (2H, *s*)  $\phi-\text{CH}_2-\text{CO}-$ ; 2.83 (1H, septet,  $J = 7$  Hz) and 1.22 (6H, *d*,  $J = 7$  Hz)  $-\text{CHMe}_2$ ; 2.56 (1H, *bs*, displaced by  $\text{D}_2\text{O}$  addition)  $-\text{OH}$ ; 1.27 (6H, *s*)  $\text{C}-\text{Me}$ . Found: C, 64.76; H, 6.78.  $\text{C}_{21}\text{H}_{26}\text{O}_7$  requires: C, 64.60; H, 6.71%.

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